## Acid Digest:

1. Each sample was mixed and sifted through a sieve
2. A $1-2 \mathrm{~g}$ sample of each soil was weighed and transferred to a 100 mL beaker
3. A $1: 1 \mathrm{HNO}_{3}$ mixture was made with 25 mL of concentrated $\mathrm{HNO}_{3}$ and 25 mL of nanopure DI water
4. 10 mL of the $\mathrm{HNO}_{3}$ mixture was added to each beaker containing a soil sample and heated until reflux for $10-15$ minutes (at $95^{\circ} \mathrm{C}+/-5^{\circ} \mathrm{C}$ ) with a watch glass over each beaker
5. The sample was taken of heating and cooling. Once cooled, 5 mL of concentrated $\mathrm{HNO}_{3}$ was added to each beaker and refluxed for 30 minutes
6. If brown fumes were observed after 30 minutes of reflux, more concentrated $\mathrm{HNO}_{3}$ was added to the reaction until no brown fumes are observed
7. The solution was refluxed for 2 hours with a watch glass at $95^{\circ} \mathrm{C}+/-5^{\circ} \mathrm{C}$
8. After 2 hours of reflux, the solution was removed from heat and cooled, and 2 mL of DI water and 3 mL of $30 \% \mathrm{H}_{2} \mathrm{O}_{2}$ were added to each reaction
9. The beakers were then returned to heat and $30 \% \mathrm{H}_{2} \mathrm{O}_{2}$ was added in 1 mL aliquots until effervescence was minimal (no more than 10 mL of $30 \% \mathrm{H}_{2} \mathrm{O}_{2}$ was added)
10. The reaction was then cooled and diluted with water in a 100 mL volumetric flask after being filtered through filter paper.
